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NONFERROUS METALS INDUSTRY STANDARD

OF THE PEOPLE'S REPUBLIC OF CHINA

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YS/T 509.2-2008

Replacing YS/T 509.2-2006

**Methods for chemical analysis of spodumene and
lepidolite concentrates - Determination of rubidium
oxide and caesium oxide contents - Flame atomic
absorption spectrometric method**

锂辉石、锂云母精矿化学分析方法

氧化铷、氧化铯量的测定

火焰原子吸收光谱法

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Table of Contents

Foreword.....	3
1 Scope.....	5
2 Method summary	5
3 Reagents	5
4 Instruments	6
5 Specimen	6
6 Analysis steps	7
7 Calculation of analysis result.....	8
8 Precision	8
9 Quality assurance and control.....	9

Foreword

YS/T 509 "Methods for chemical analysis of spodumene and lepidolite concentrates" consists of the following 11 parts:

- Part 1: Determination of lithium oxide sodium oxide and potassium oxide contents - Flame atomic absorption spectrometric method;
- Part 2: Determination of rubidium oxide and caesium oxide contents - Flame atomic absorption spectrometric method;
- Part 3: Determination of silicon dioxide content - Gravimetric-molybdenum blue photometric method;
- Part 4: Determination of aluminum oxide content - EDTA complexometric method;
- Part 5: Determination of ferric oxide content-0-phenanthroline photometric method and EDTA complexometric method;
- Part 6: Determination of phosphorus pentoxide content-The molybdenum blue photometric method;
- Part 7: Determination of beryllium oxide content – Chromazurol S and cetyltrimethyl ammonium bromide spectrophotometric met;
- Part 8: Determination of calcium oxide and magnesium oxide content - Flame atomic absorption spectrometric method;
- Part 9: Determination of fluorine content - Ion selective electrode method;
- Part 10: Determination of manganous oxide content-Persulfate oxidation photometric method;
- Part 11: Determination of loss on ignition - Gravimetric method.

This Part is Part 2.

This Part replaces YS/T 509.2-2006 "Methods for chemical analysis of spodumene and lepidolite concentrates The atomic absorption spectrophotometric method for determination of rubidium oxide and caesium oxide contents" (previous edition is GB/T 3885.2-1983).

Compared with YS/T 509.2-2006, the main changes in this Part are as follows:

- re-edited the text format;
- added the provisions on quality assurance and control; added the

Methods for chemical analysis of spodumene and lepidolite concentrates - Determination of rubidium oxide and caesium oxide contents - Flame atomic absorption spectrometric method

1 Scope

This Part specifies the determination methods for rubidium oxide and caesium oxide contents in spodumene and lepidolite concentrates.

This Part is applicable to the determinations of rubidium oxide and caesium oxide contents in spodumene and lepidolite concentrates. Determination range: 0.50%~3.00% for rubidium oxide; 0.10%~1.20% for caesium oxide.

2 Method summary

The test material is decomposed with hydrofluoric acid-sulfuric acid. Add a certain amount of potassium sulfate as a deionizer. In sulfuric acid medium, at the atomic absorption spectrometer wavelengths of 780.0nm and 852.1nm, use air-acetylene lean-burn flame to respectively determine rubidium oxide, caesium oxide.

3 Reagents

Unless otherwise stated, only the confirmed analytically-pure reagents and distilled water or water of comparable purity are used in the analysis.

3.1 Hydrofluoric acid (p1.15g/mL).

3.2 Sulfuric acid (1+1).

3.3 Potassium sulfate solution: Weigh 37.0g of potassium sulfate. Put it in a 250mL beaker. Add water to dissolve. Transfer to a 1000mL volumetric flask. Use water to dilute to the graduation of scale. Mix well. 1mL of this solution contains 20mg of potassium oxide.

3.4 Rubidium oxide standard stock solution: Weigh 1.2937g of rubidium chloride (above 99.9%) that has been baked at 105°C~110°C for 2h and placed

0.074mm. The lepidolite concentrate shall pass through a sieve with an aperture of 0.105mm.

5.2 The specimen is pre-baked at 105°C~110°C for 2h. Place in a desiccator to cool to room temperature.

6 Analysis steps

6.1 Test material

Weigh 0.25g of specimen, to the nearest of 0.0001g.

6.2 Number of determinations

Make parallel determinations of two test materials. Take the average value.

6.3 Blank test

Do a blank test with the test material.

6.4 Determination

6.4.1 Place the test material (6.1) in a 50mL platinum dish. Use water to moisture. Add 3mL of sulfuric acid (3.2), 15mL of hydrofluoric acid (3.1). Put it on the sand bath and heat it to decompose. Steam to emit thick white smoke of sulfur trioxide. Remove. Cool.

6.4.2 Add 20mL~25mL of water. Put it on a sand bath and heat it to completely dissolve the salt. Remove. Cool to room temperature.

6.4.3 Transfer the solution (6.4.2) into a 100mL volumetric flask. Add 10.0mL of potassium sulfate solution (3.3). Use water to dilute to the graduation of scale. Mix well.

6.4.4 Pipette 20.00mL of solution (6.4.3). Place in a 50mL volumetric flask. Add 0.6mL of sulfuric acid (3.2), 3.0mL of potassium sulfate solution (3.3). Use water to dilute to the graduation of scale. Mix well.

6.4.5 Put the solution (6.4.4) on the atomic absorption spectrometer. At a wavelength of 780.0nm, with air-acetylene lean-burn flame, with water for zero-adjustment, measure the absorbance of rubidium oxide. Find out the corresponding amount of rubidium oxide from the working curve.

6.4.6 Put the solution (6.4.3) on the atomic absorption spectrometer. At a wavelength of 852.1nm, with air-acetylene lean-burn flame, with water for zero-adjustment, measure the absorbance of caesium oxide. Find out the corresponding amount of caesium oxide from the working curve.

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